Distinguish, differentiate, compare and explain what is the main difference between Ionic and Covalent compounds. Comparison and Differences.

Difference between Ionic and Covalent compounds

- 1. In term of strength, ionic compounds have strong forces of attraction whereas covalent compounds have weak forces of attraction.
- 2. Ionic compounds are soluble in water. Covalent compounds are insoluble in water.
- 3. Ionic compounds conduct electricity in aqueous solutions. Covalent compounds do not conduct electricity in aqueous solution.

